



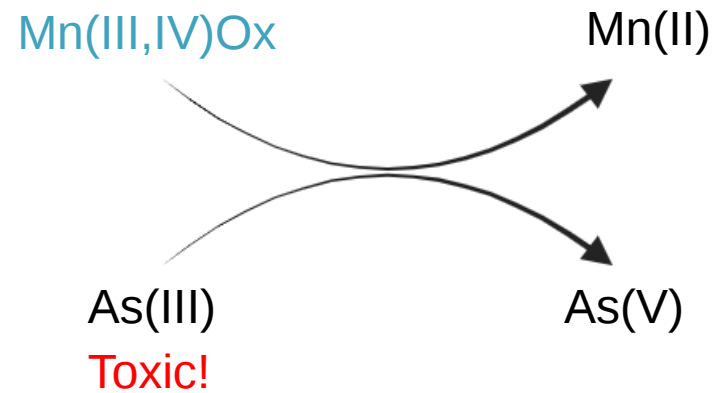
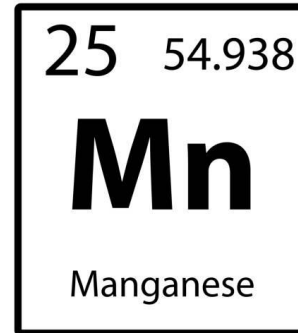
UC San Diego



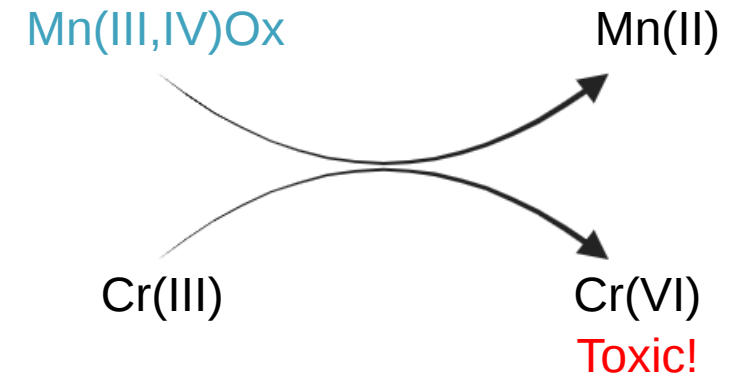
Microbial biofilms structure and manganese (bio)cycling: from lab to field studies

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J.M. Diaz – Scripps Inst., A. Gélabert – IPGP

Manganese is a central element in biogeosciences

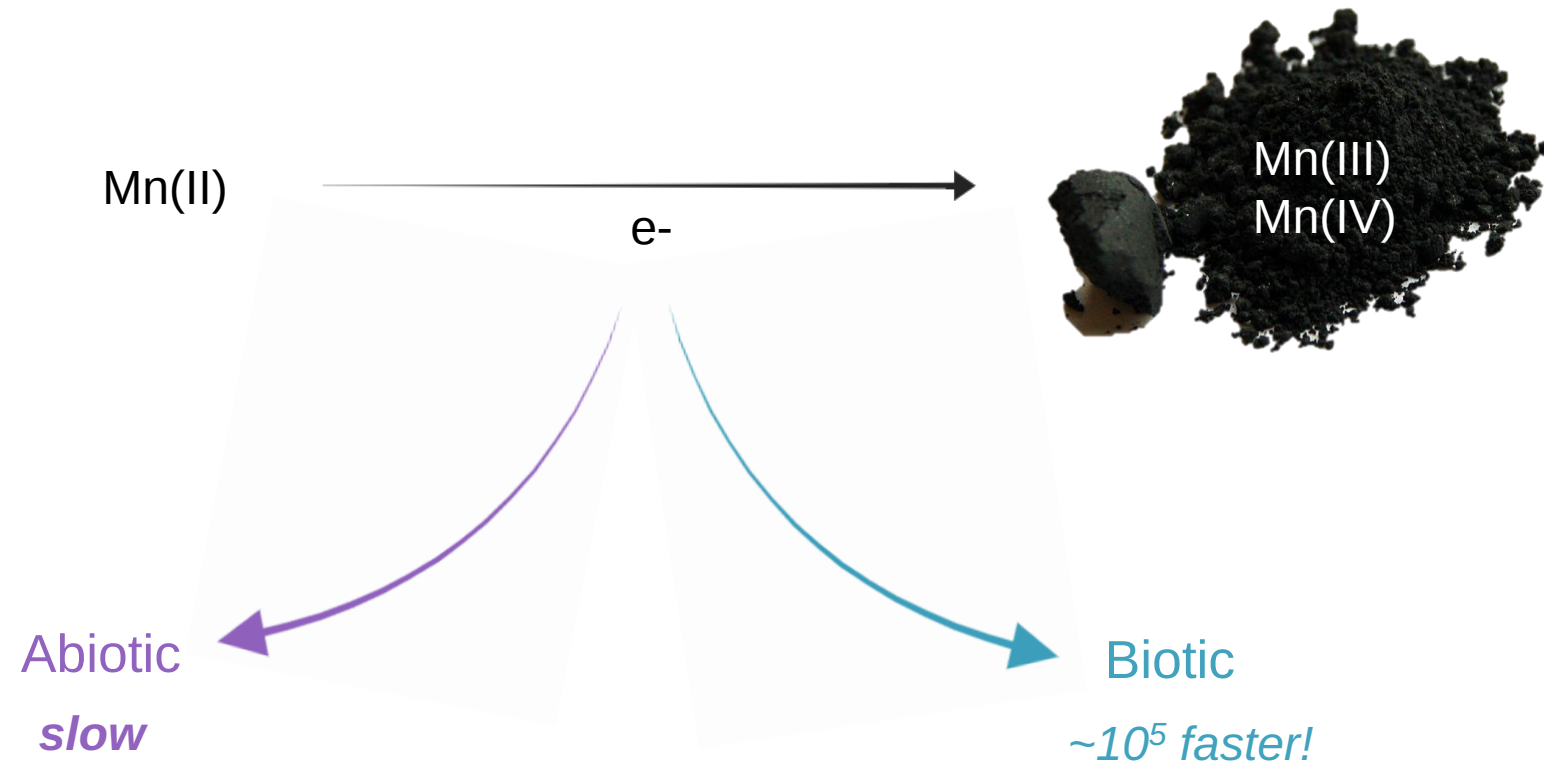


(PM. H, D. L. Sparks et al., 1991)
(Tebo et al, 2004)



(Butler et al, 2015)
(Tebo et al, 2004)

Two ways to form manganese oxides in the environment



(Diem and Stumm, 1984)

(Nealson, Tebo and Rosson, 1988)

Two ways to form manganese oxides **biotically**

Direct

Enzyme



Two ways to form manganese oxides biotically

Direct

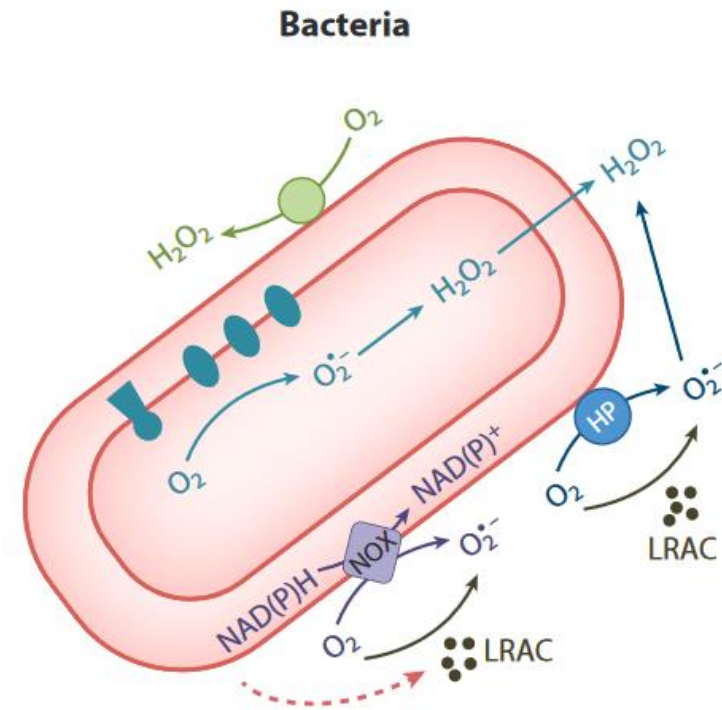
Enzyme



(Tebo BM et al, 2005)

Indirect

Reactive Oxygen Species (ROS)



(C. Hansel and J.M. Diaz, 2021)
(Learman et al,⁵2011)

Two ways to form manganese oxides biotically

Direct

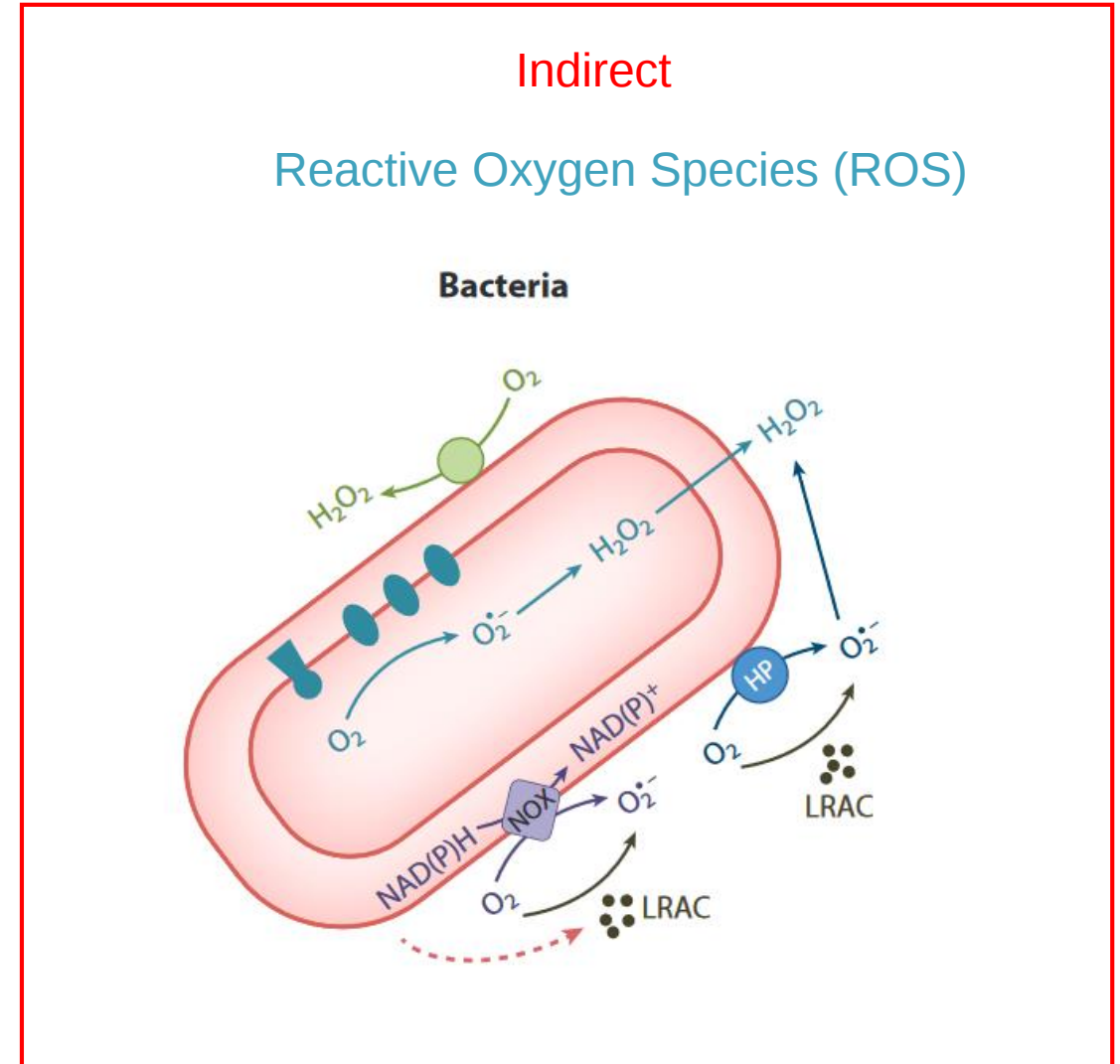
Enzyme



(Tebo BM et al, 2005)

Indirect

Reactive Oxygen Species (ROS)

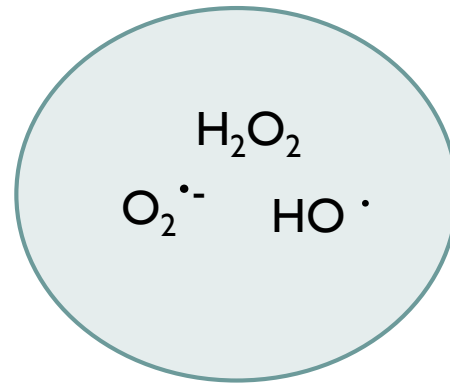


(C. Hansel and J.M. Diaz, 2021)

(Learman et al, 2011)

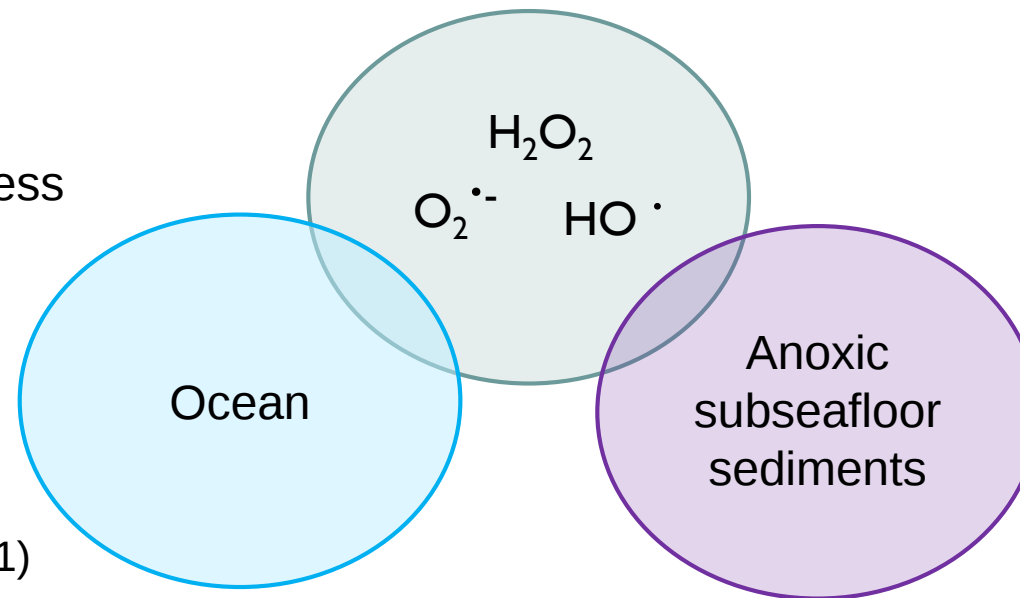
Microorganisms release reactive oxygen species (ROS) (Diaz. et al., *Science*, 2013)

- Transient
- Highly reactive species
- +++ oxidizers
- Key but cryptic process



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(C. Hansel and J. M. Diaz, 2021)
(Sutherland et al, 2020)

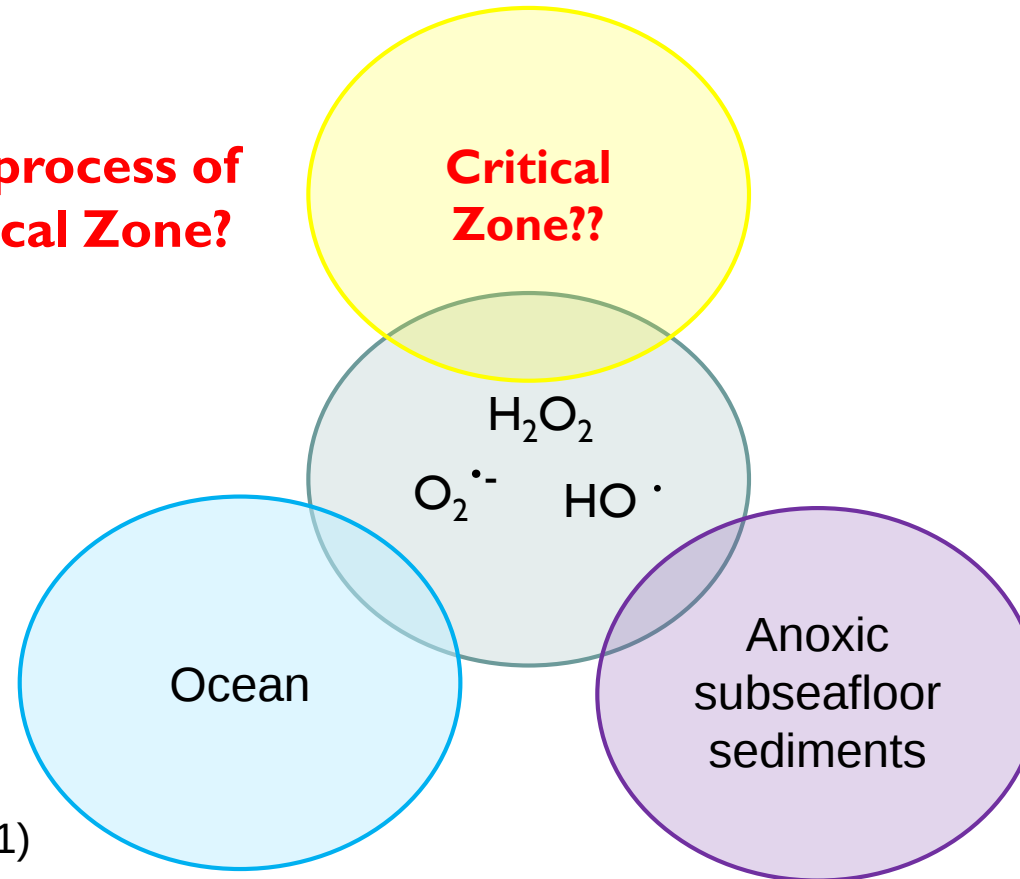
Sink of marine dissolved oxygen
~15 – 50 % photosynth. O_2

(Sauvage et al, 2021)
(A. I. Garber et al, 2021)

Neutralization of water radiolysis ROS
by microorganisms may provide an *in situ* source of oxygen

Microorganisms release reactive oxygen species (ROS) (Diaz. et al., *Science*, 2013)

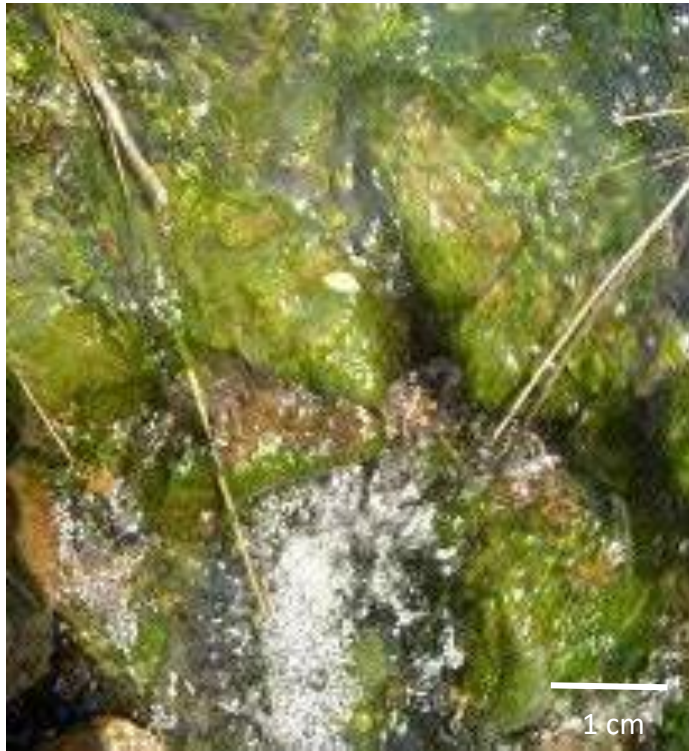
**Extracellular ROS = a key process of
Mn (bio)cycling in the Critical Zone??**



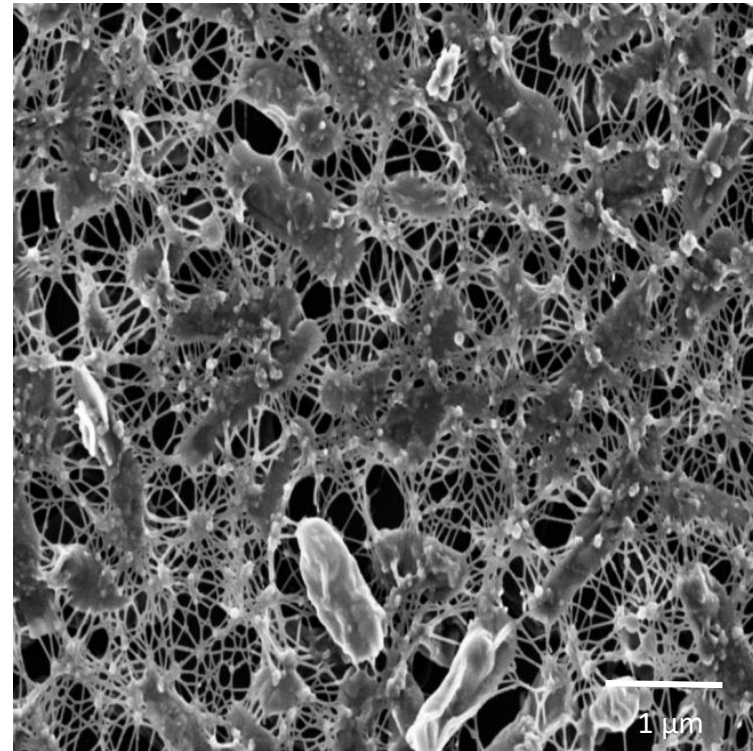
(C. Hansel and J. M. Diaz, 2021)
(Sutherland et al, 2020)

(Sauvage et al, 2021)
(A. I. Garber et al, 2021)

Role of microbial biofilms

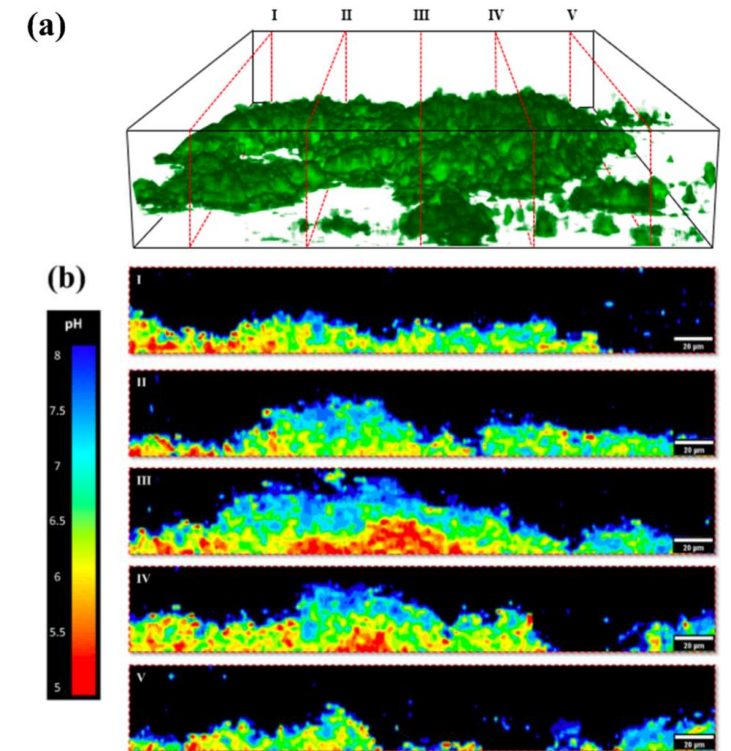


Gel-like structure



©T. Couasnon

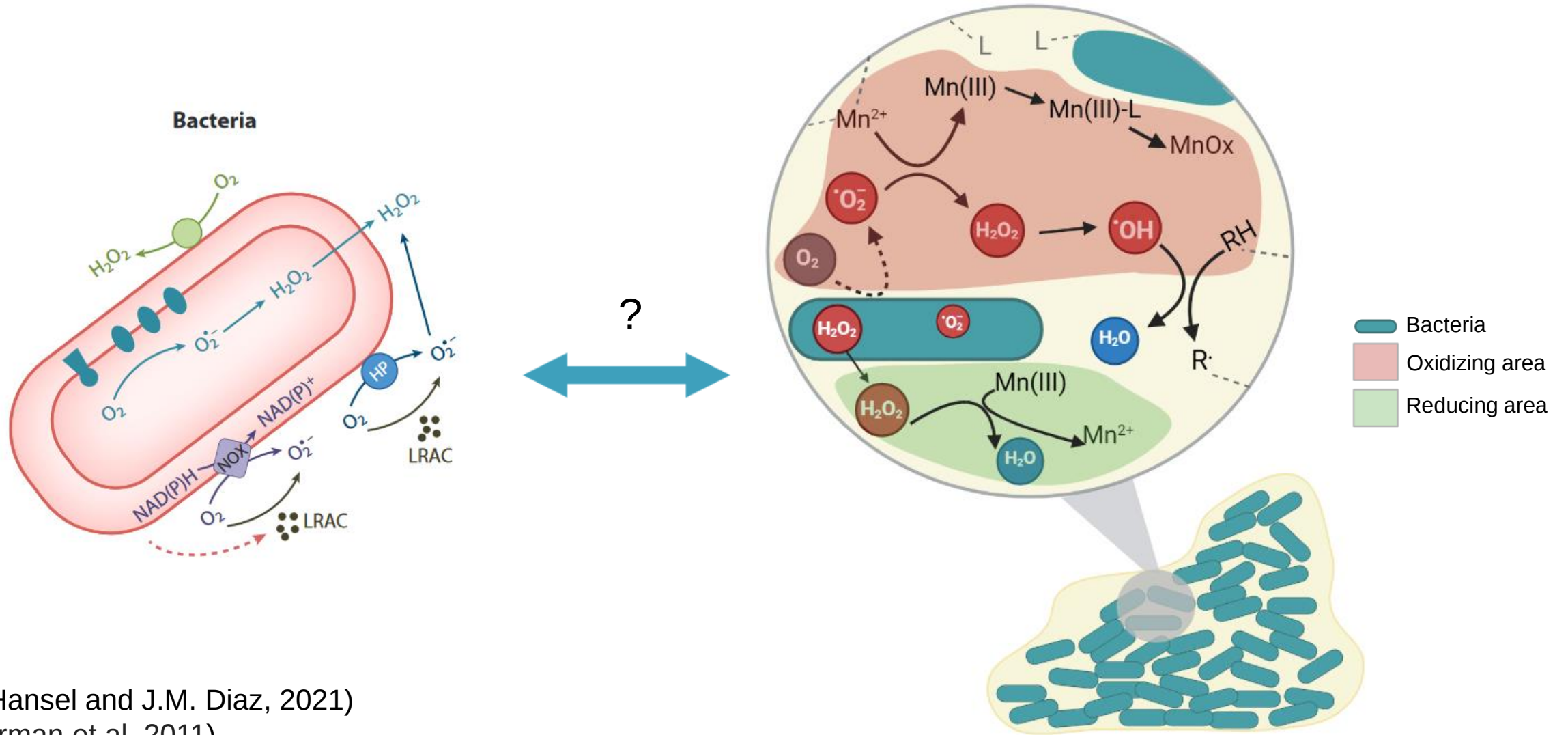
Limited transport



μ-environments

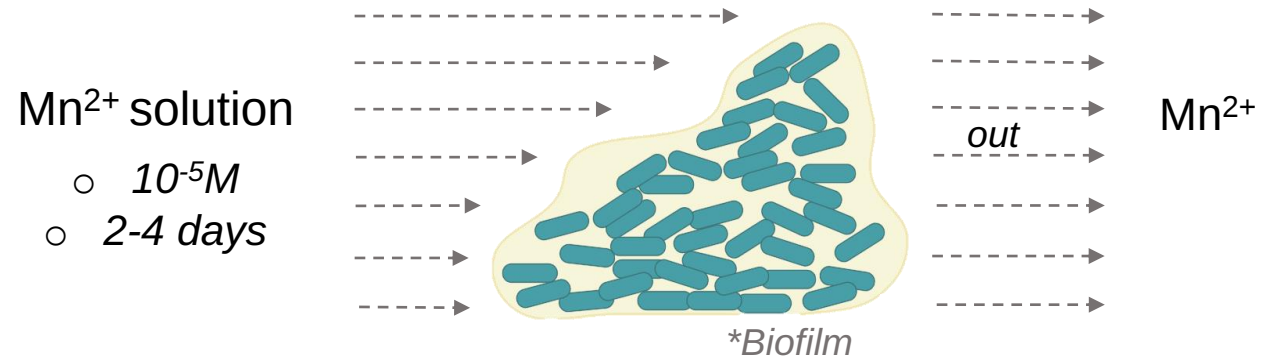
(Fulaz et al, 2019)

Biofilms accumulate ROS creating highly oxidizing pockets



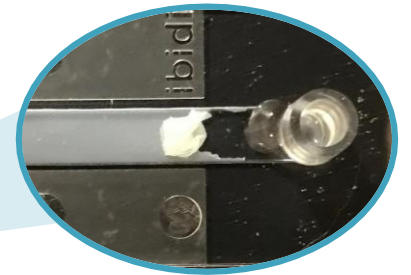
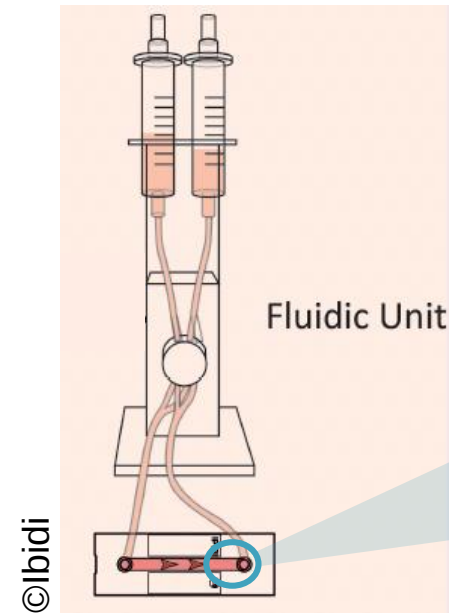
(C. Hansel and J.M. Diaz, 2021)
(Learman et al, 2011)

Laboratory experimental setup

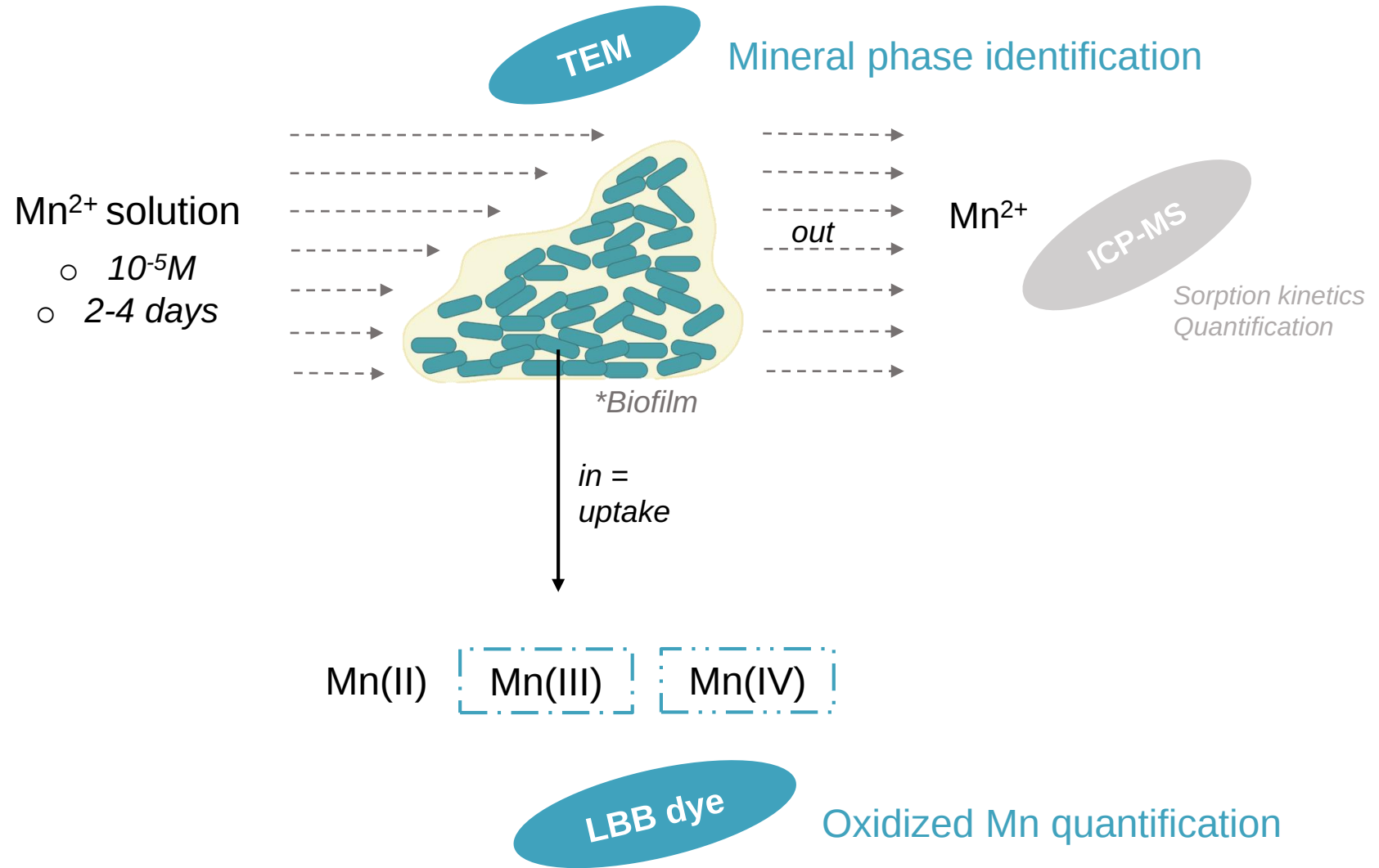


*Strain used *E. coli* K12 MG1655 F'tet

- Controlled system
- No Mn-oxidation related enzyme

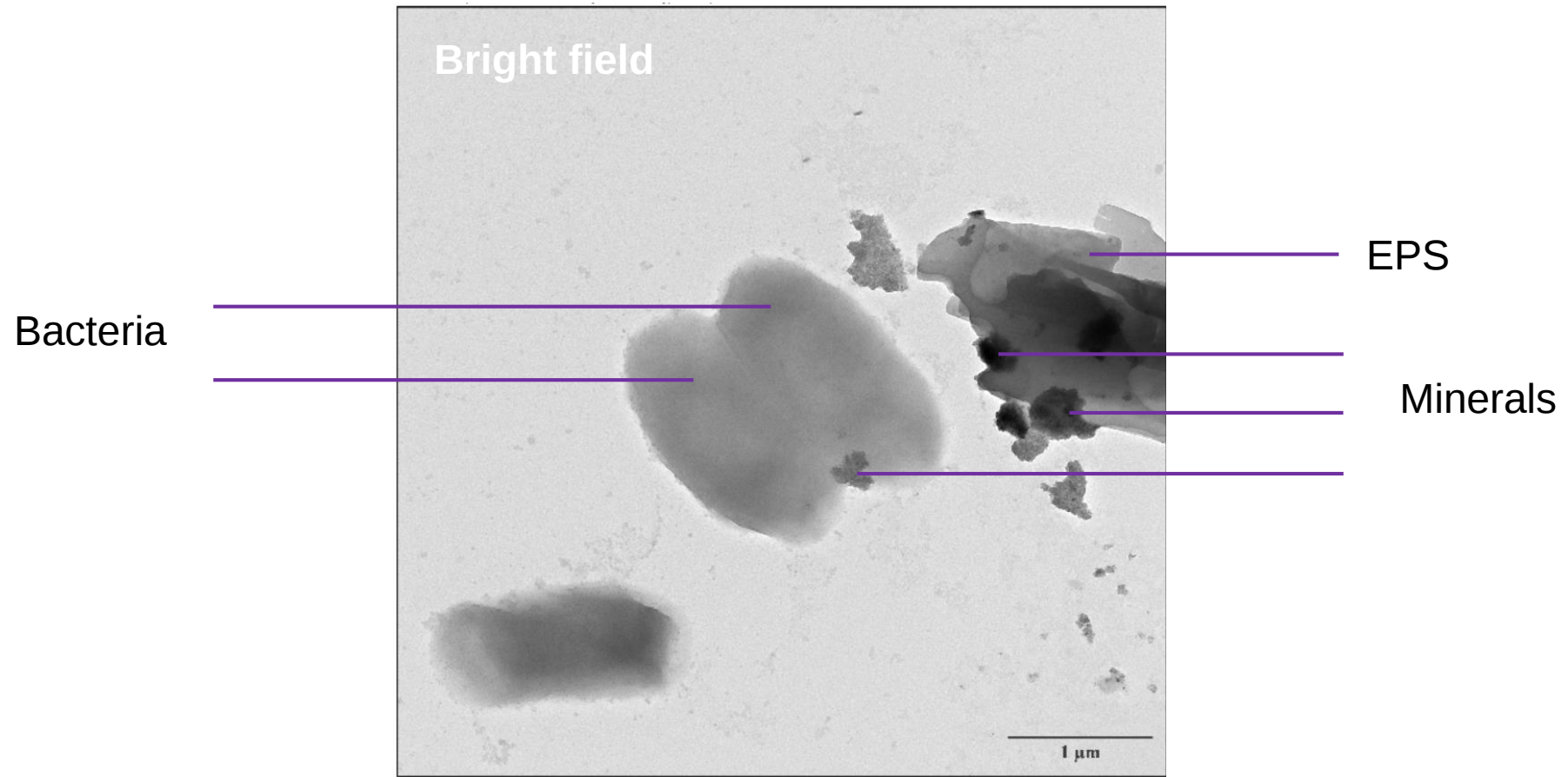


Laboratory experimental setup

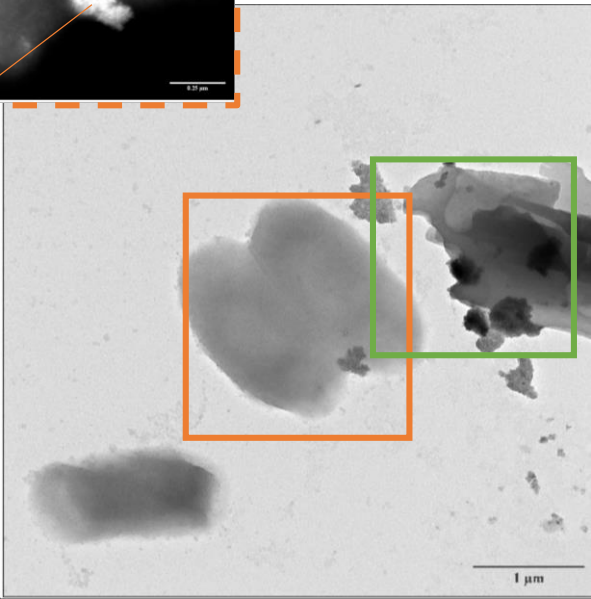
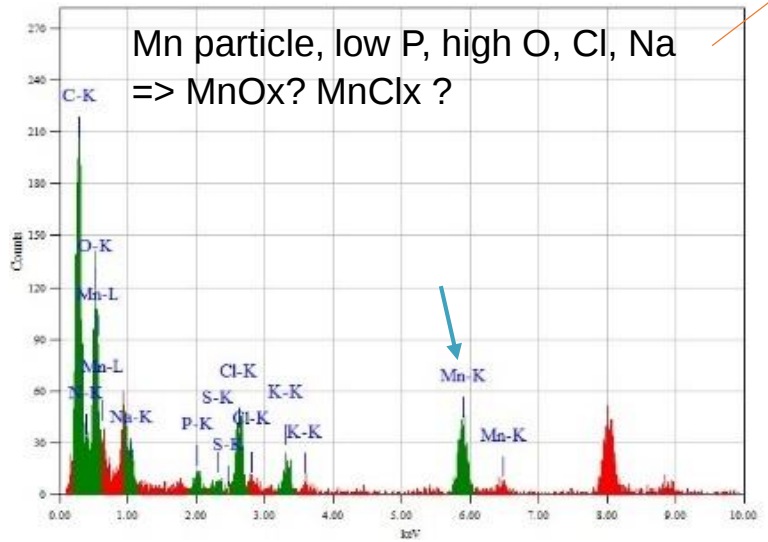
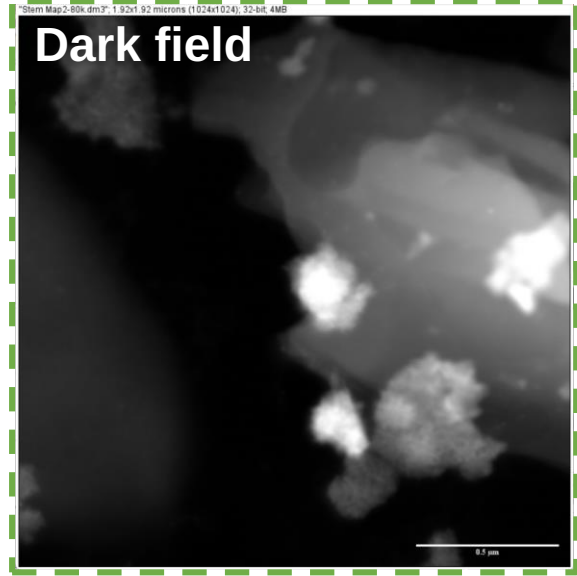
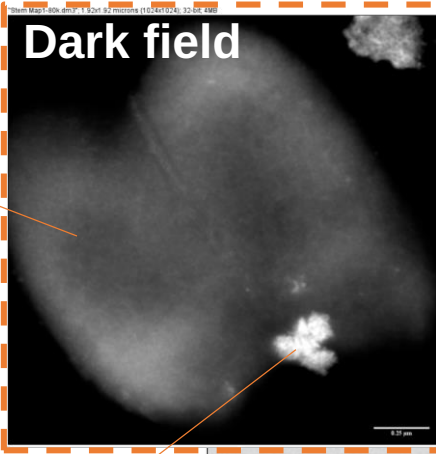
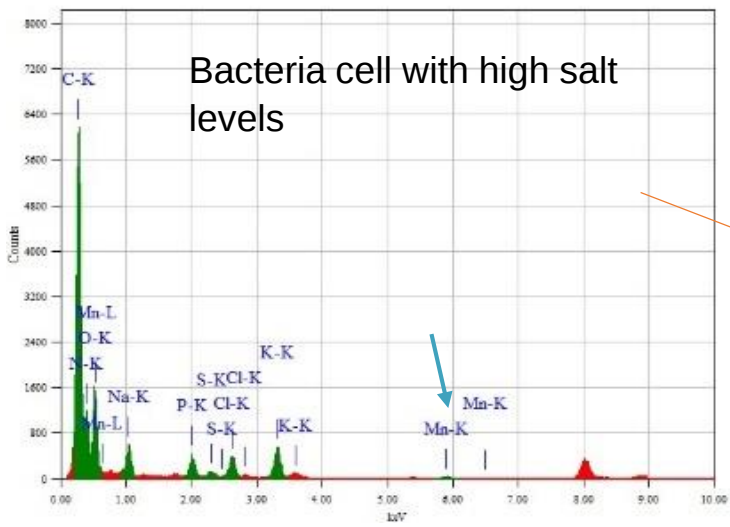


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Mineral phase identification - TEM

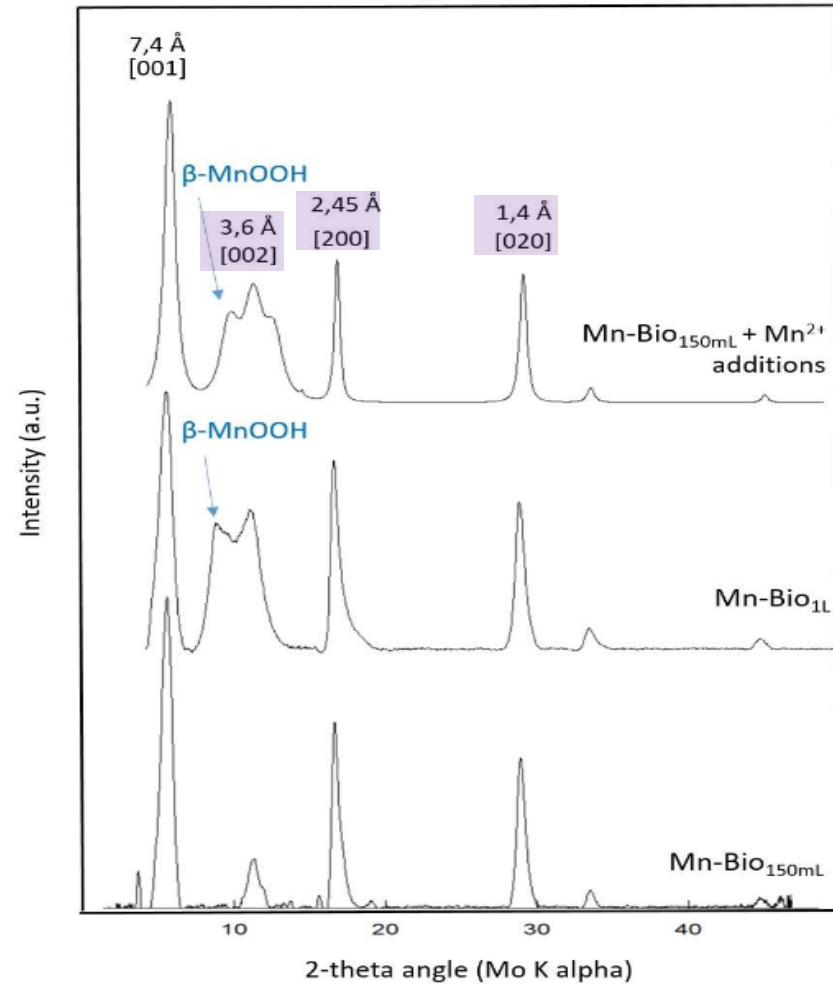
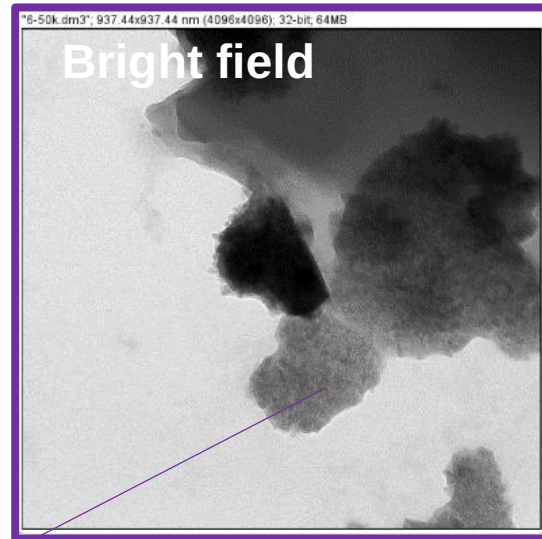
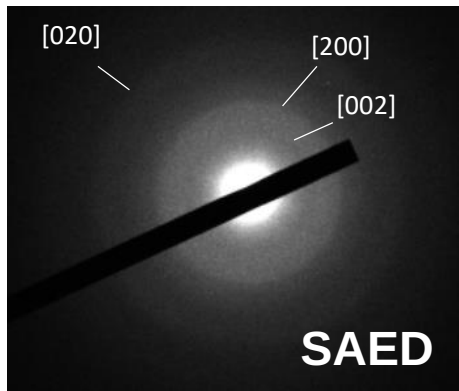


Mineral phase identification - TEM



Mineral phase identification - TEM

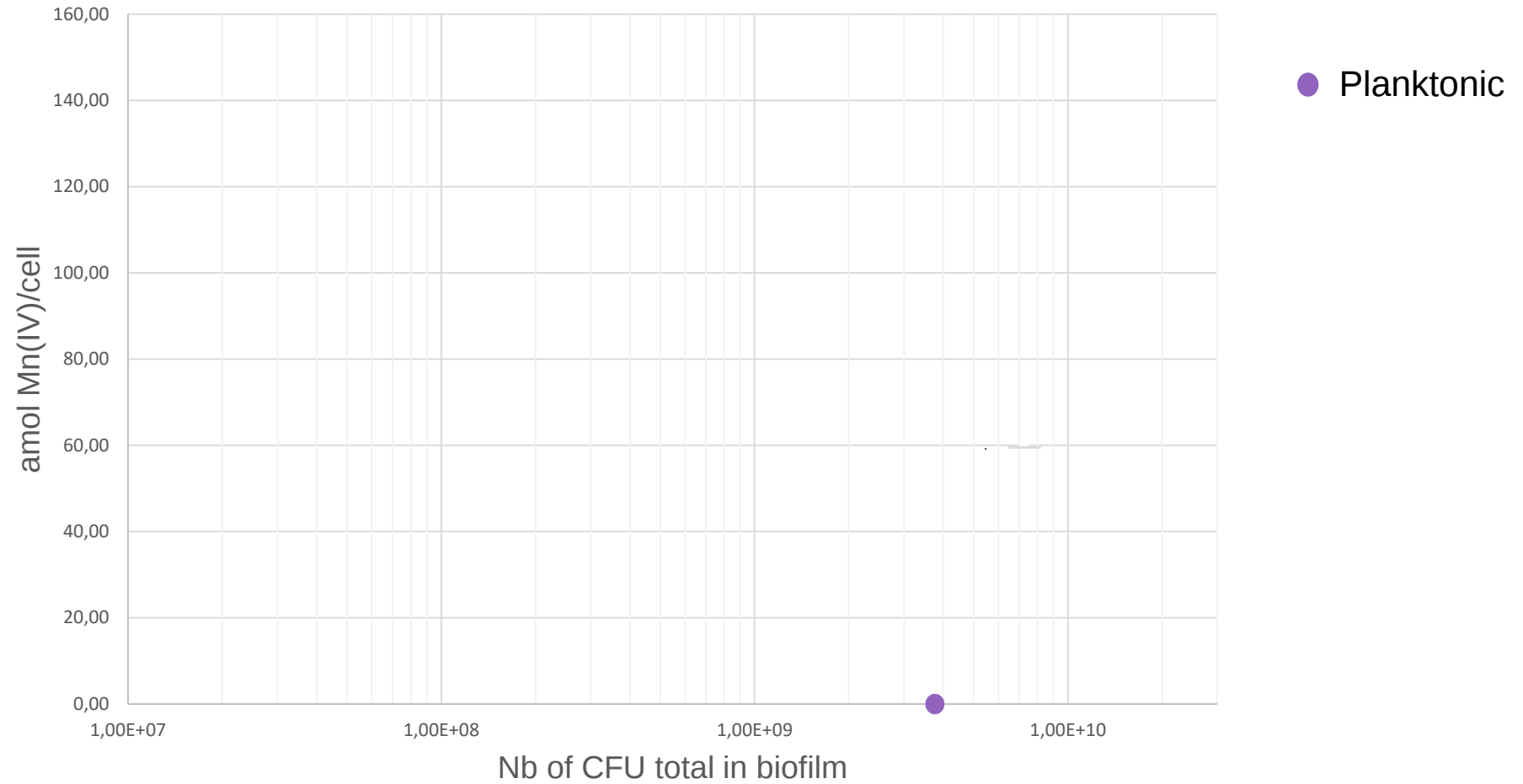
Minerals are related to a **BIRNESSITE** structure



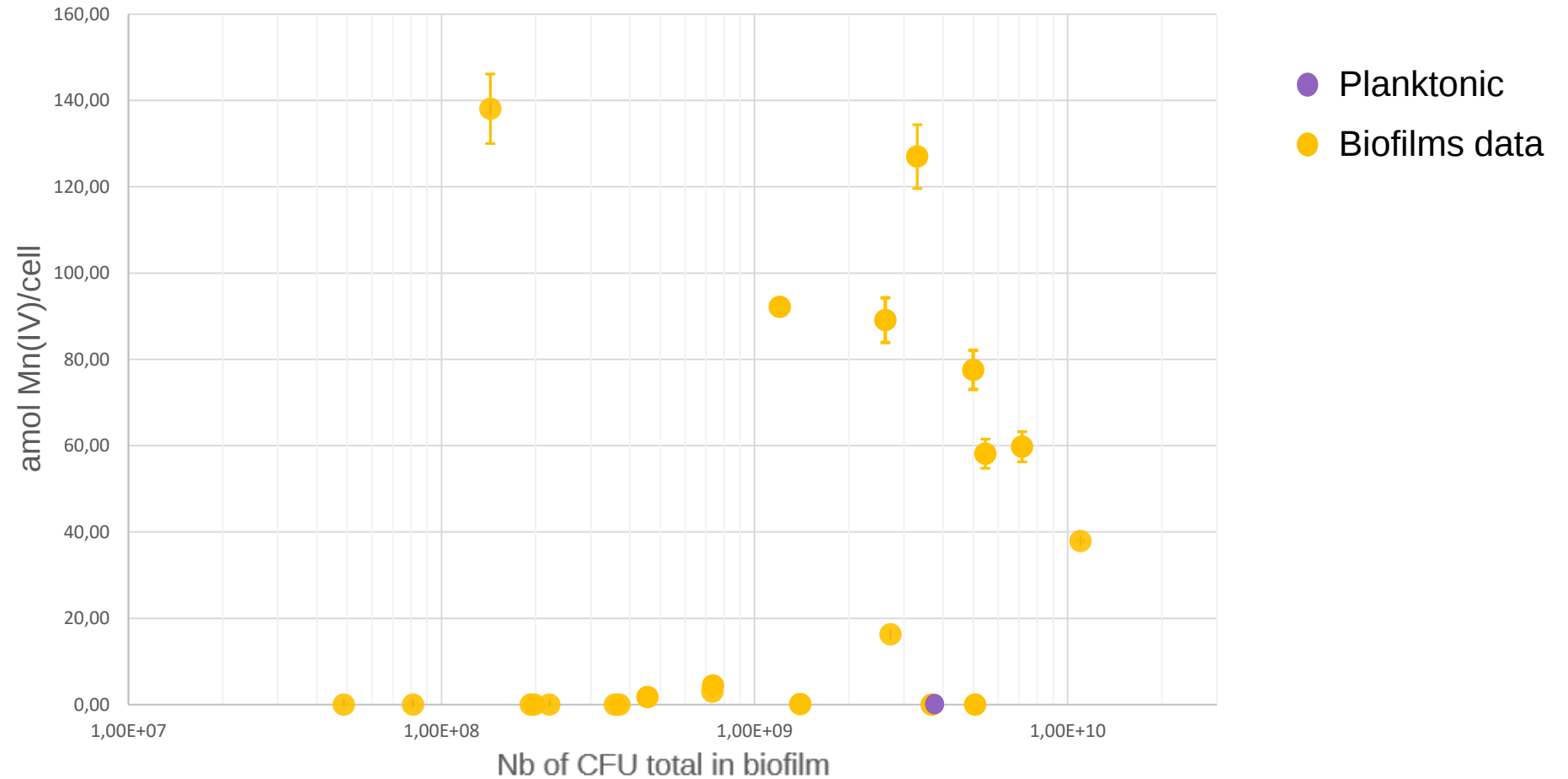
XRD analysis of birnessite biomineral, (L. Galezowski, 2021)

Evidence of Mn oxidation in the biofilm

Oxidized Mn quantification – LBB dye



Oxidized Mn quantification – LBB dye

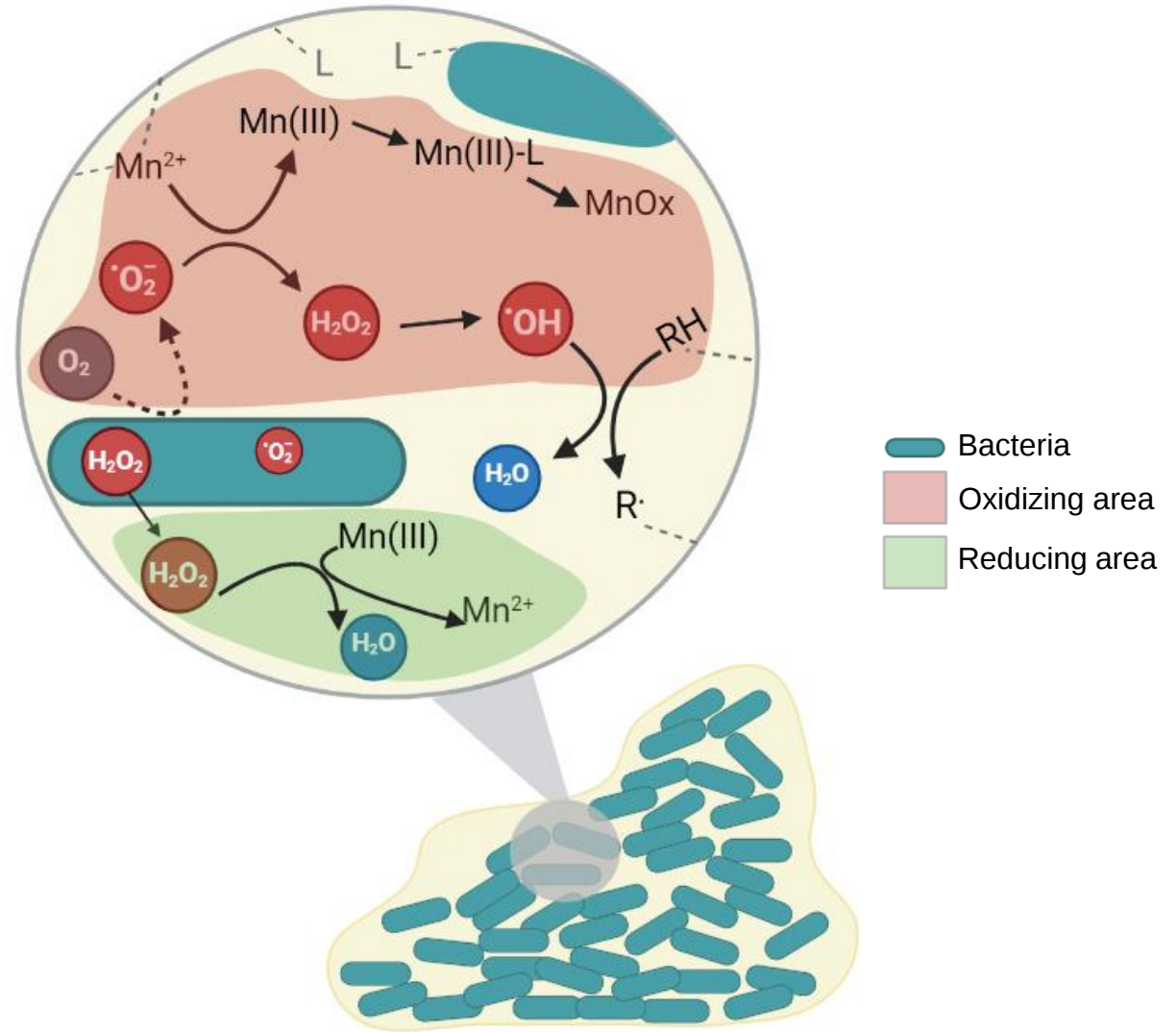


E. coli biofilms are able to oxidize Mn(II)!

Biofilms accumulate ROS creating highly oxidizing pockets

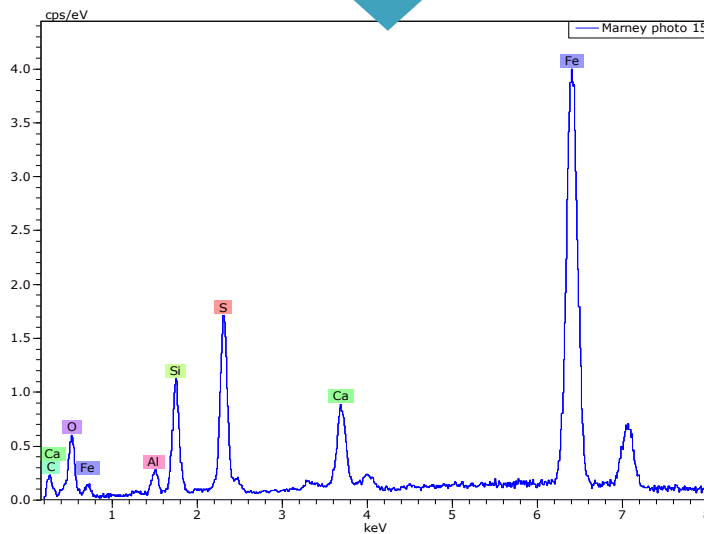
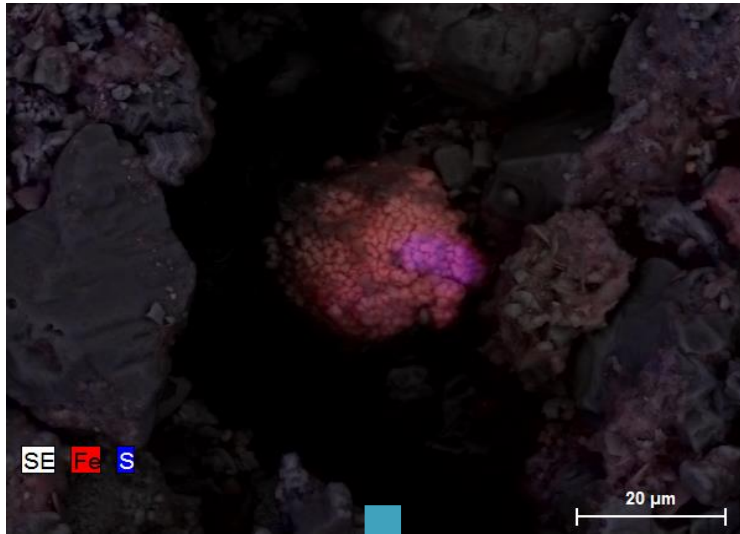
Work in progress:

- CLSM imaging targeting ROS
- Field sampling

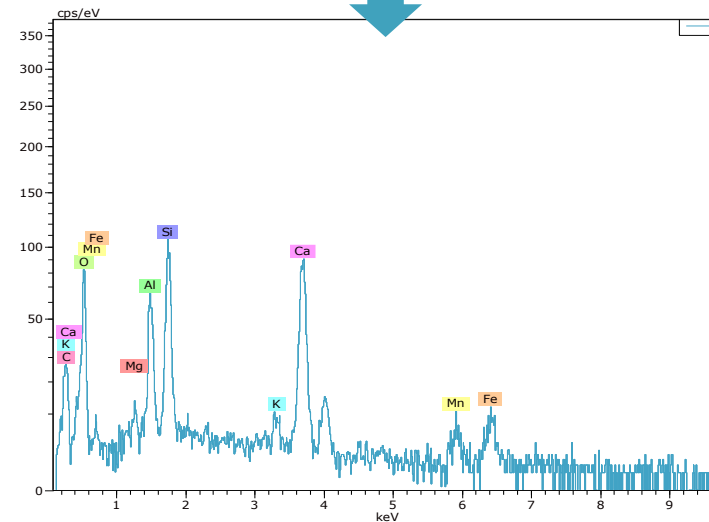
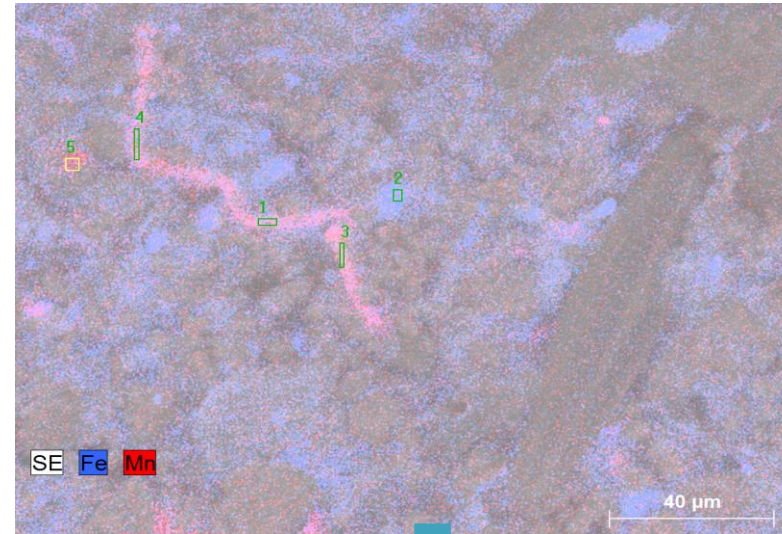


Microenvironments in natural biofilms (Seine river)

Co auteurs:
A. Gelabert, T. Berthe, G. Morin, Y. Colin, F. Petit



Framboidal pyrite
Reducing environments



Mn oxides formed *in situ*?
Oxidative environments?

Take home message

- Biofilms seem to get specific redox microenvironments
- *E. coli* biofilms are able to oxidize Mn(II)
- Biological (and abiotic) ROS production in the critical zone needs further investigation -> **Major role for redox-sensitive elements?**

Thank you

François Guyot
Bénédicte Menez
Stephan Borensztajn
Céline Pisapia
Léna Lecourt
Emmanuelle Gérard
Jean-Marc Ghigo

